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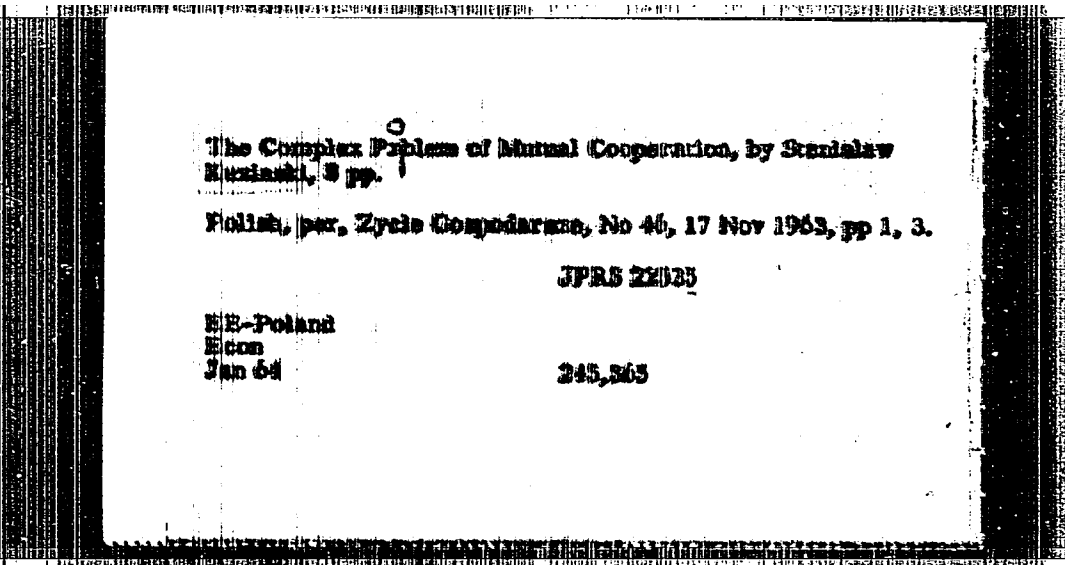

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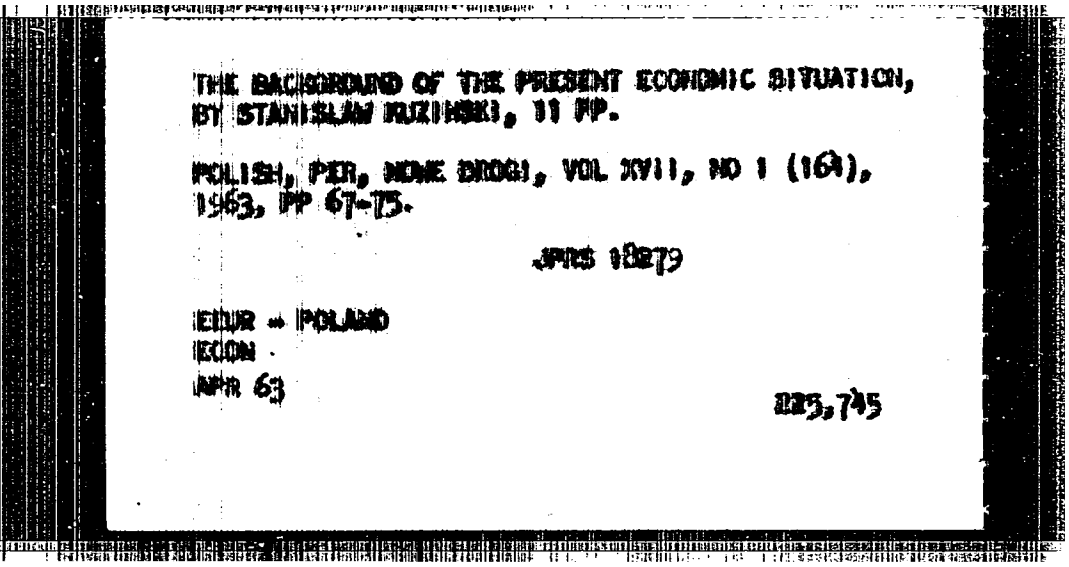

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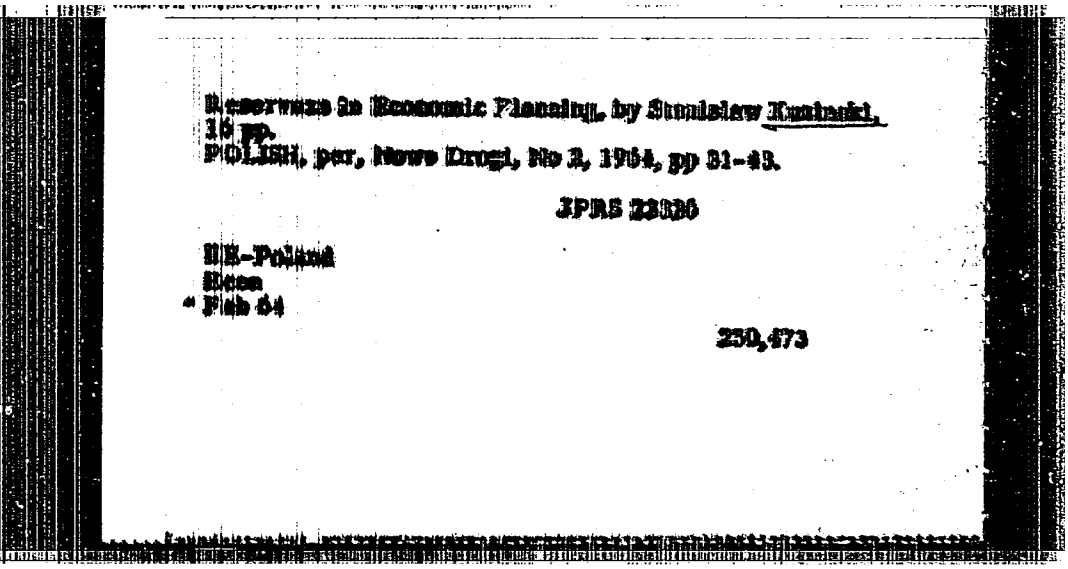

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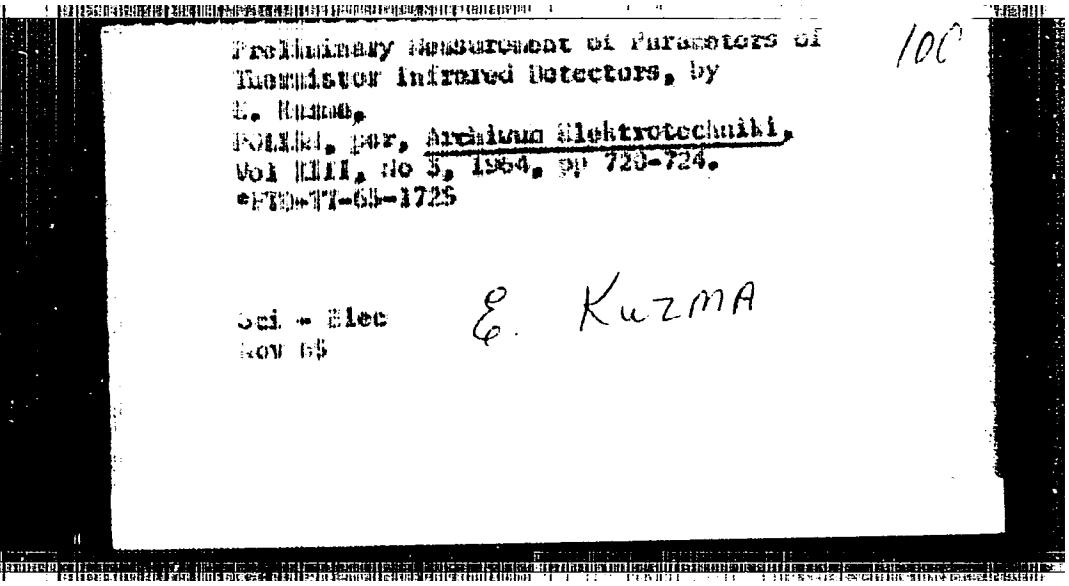

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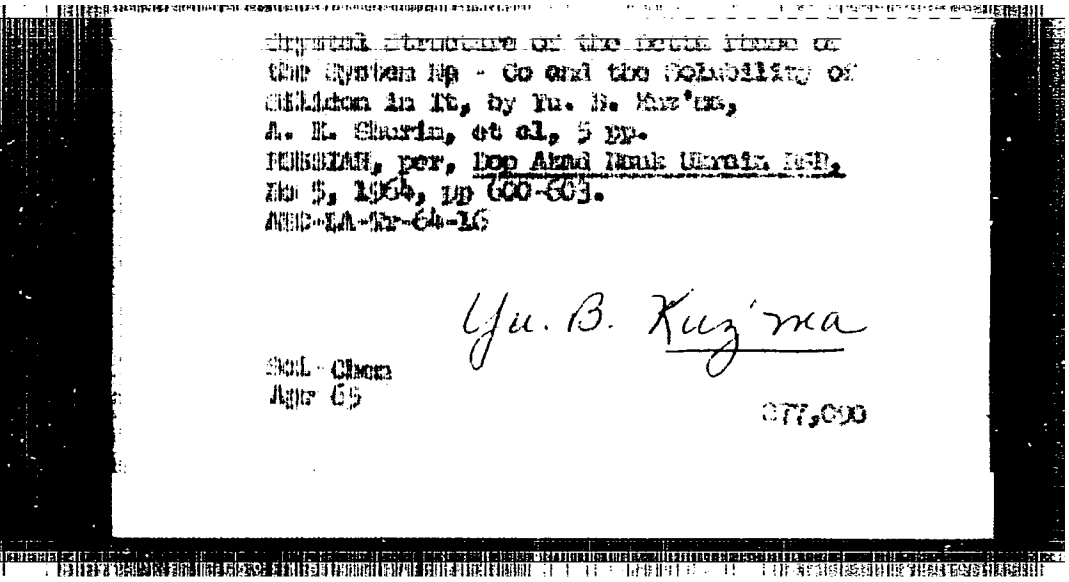

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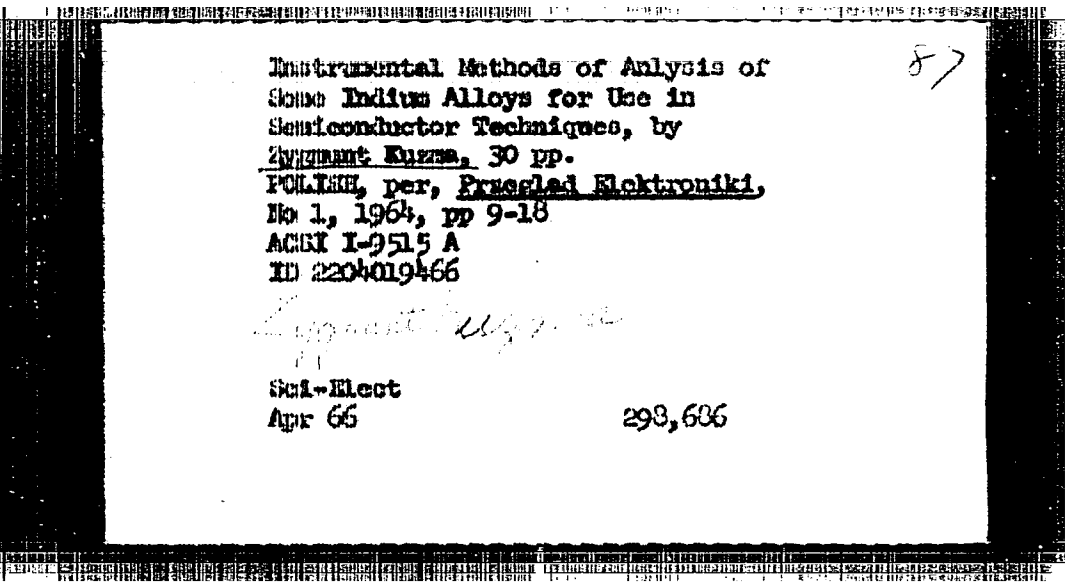

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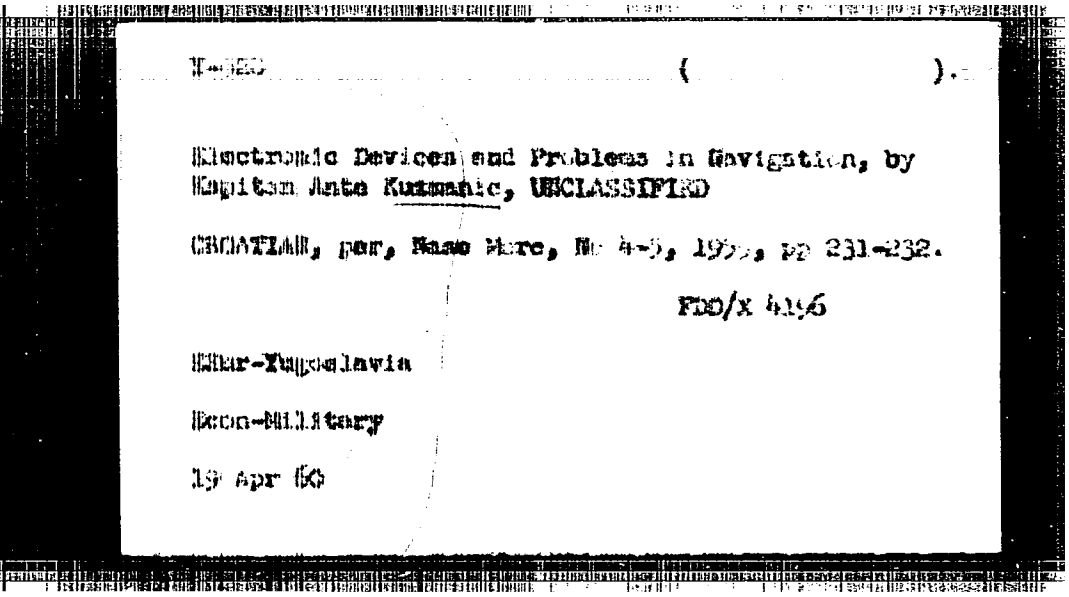

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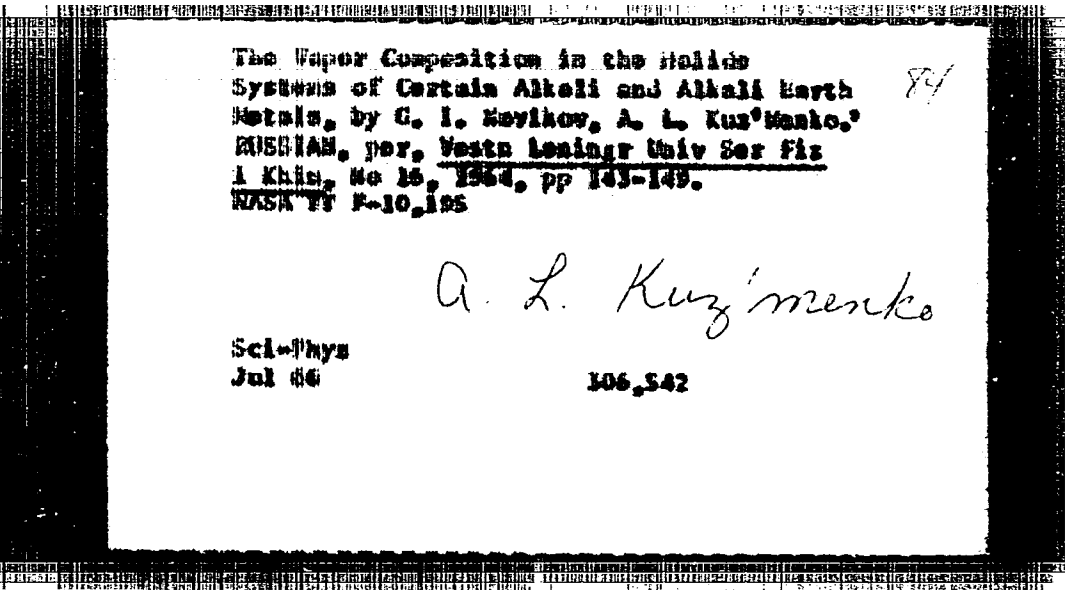

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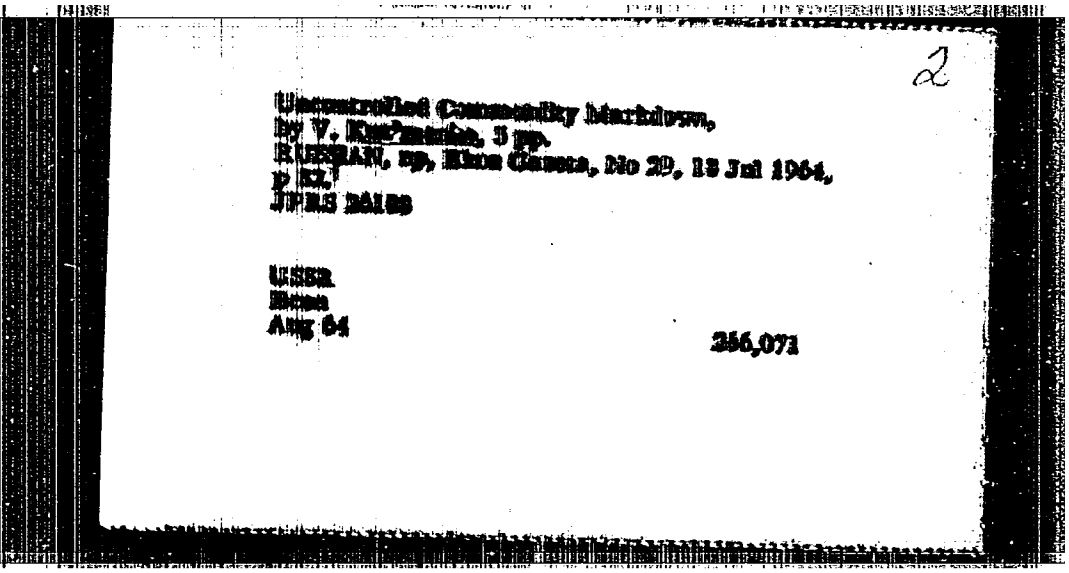

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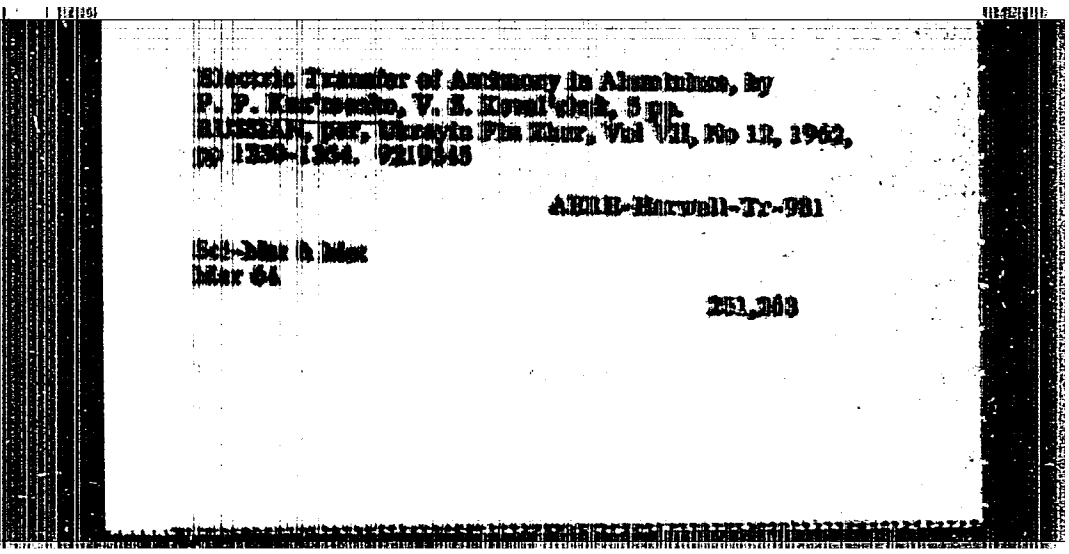

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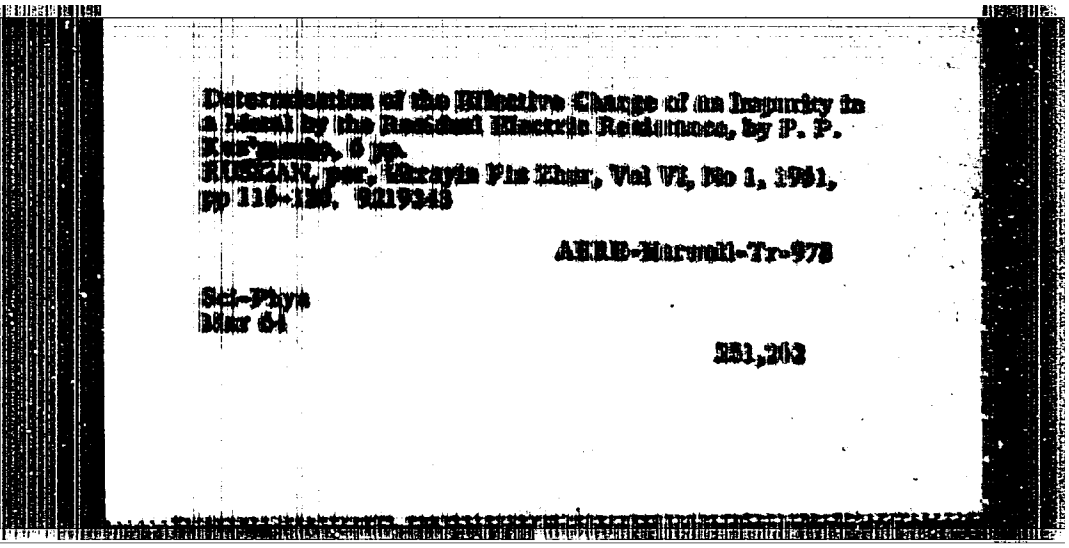

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The expression

$$v = \frac{U_{mn}}{KT}\left(ql - \frac{\delta}{\sigma} i^2\right) = \frac{1}{k\sigma_0}(ql - \rho i^2)D$$

is derived for the mean velocity (v) of an atom, in a crystal lattice, that is acted on by (1) the electric field produced by passing a current through the metal and (2) the collision of conduction electrons through the crystal lattice. Here q is the charge on the ion, δ is the unit shift, i is current density, σ is specific conductivity, D is the diffusion coefficient in the
(Physics--Solid State, TT, v. 5, no. 9) (over)

61-15307

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2. Ions--Motion
3. Silver--Diffusion
4. Electric fields--Physical effects

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APPROVED FOR RELEASE: 08/15/ 2001 CIA-RDP84-00581R000100310030-0